

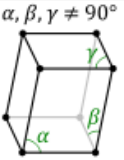
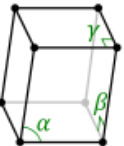
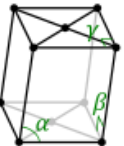
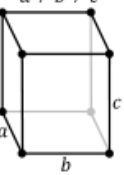
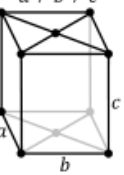
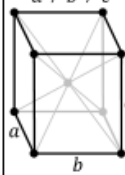
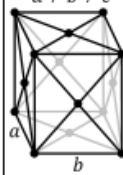
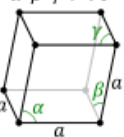
# Miller Indices

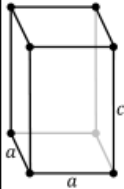
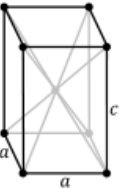
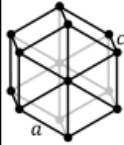
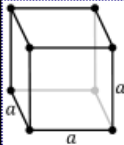
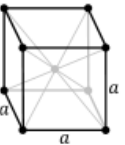
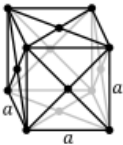
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Ahmad Asi

# Introduction

- Developed by William Miller – 1839
- Primarily to describe crystalline structures
- Indices used to reference orientation of planes
- Consist of positive & negative integers
- Help determine material properties

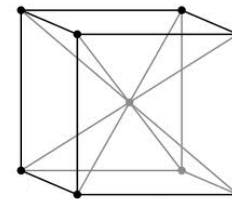
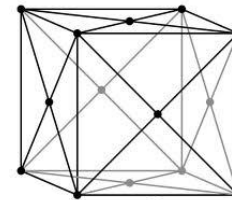
# Examples of Crystal Structures

The 7 lattice systems (From least to most symmetric)	The 14 Bravais Lattices			
1. <b>triclinic</b> (none)	$\alpha, \beta, \gamma \neq 90^\circ$ 			
2. <b>monoclinic</b> (1 diad)	simple	base-centered		
	$\alpha \neq 90^\circ$ $\beta, \gamma = 90^\circ$ 	$\alpha \neq 90^\circ$ $\beta, \gamma = 90^\circ$ 		
3. <b>orthorhombic</b> (3 perpendicular diads)	simple	base-centered	body-centered	face-centered
	$a \neq b \neq c$ 	$a \neq b \neq c$ 	$a \neq b \neq c$ 	$a \neq b \neq c$ 
	$\alpha = \beta = \gamma \neq 90^\circ$			
				

5. <b>tetragonal</b> (1 tetrad)	simple	body-centered	
	$a \neq c$ 	$a \neq c$ 	
6. <b>hexagonal</b> (1 hexad)			
7. <b>cubic</b> (4 triads)	simple (SC)	body-centered (bcc)	face-centered (fcc)
			
	$a$		

# Definitions

- Lattice Parameters
  - Lengths of the edges of a lattice, defined by distance
- Isotropic
  - Uniform in all orientations, regardless of reference point
- Anisotropic
  - Interactive properties are dependent on incident orientation
- Face-Centered Cubic
  - Cubic lattice with atoms on each face
- Body-Centered Cubic
  - Cubic lattice with atom in center



# Nomenclature

- Indices expressed as integers
- x, y, z axis correspond to lengths a, b, c
- a, b, c, are inverted to find h, k, l
- Negatives expressed with 'bar':  $\bar{1}$ ,  $\bar{2}$ ,  $\bar{3}$ , etc.
- ( ) Brackets represent single planes
- { } Brackets represent family of planes

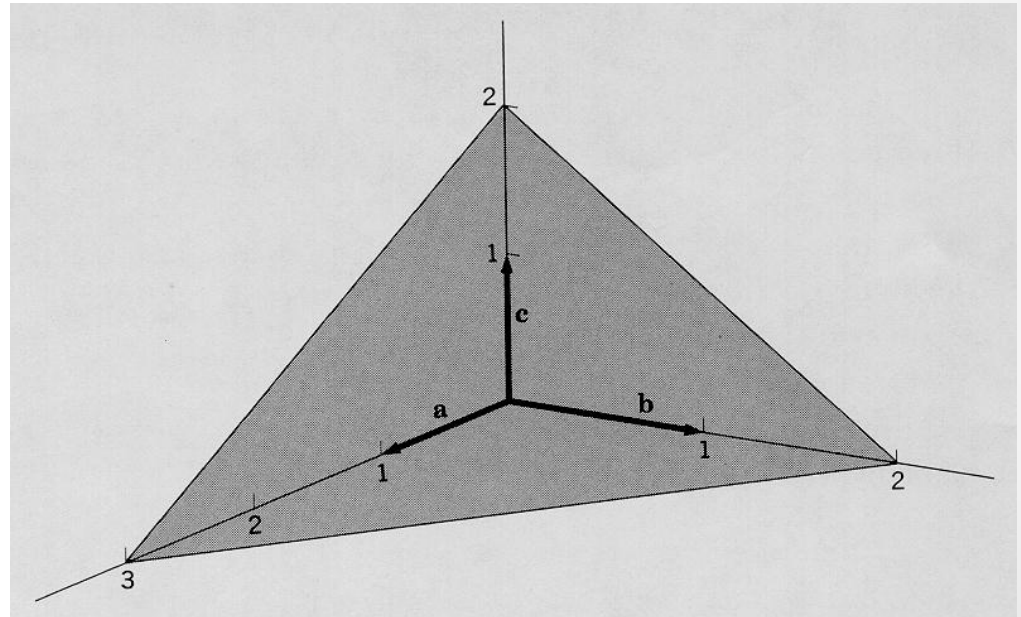
# Determine Ratios

- Evaluate axis intercepts
  - Equal to  $1/h$ ,  $1/k$ ,  $1/l$
- Invert to find:
  - $h, k, l$

$$h = 1/3$$

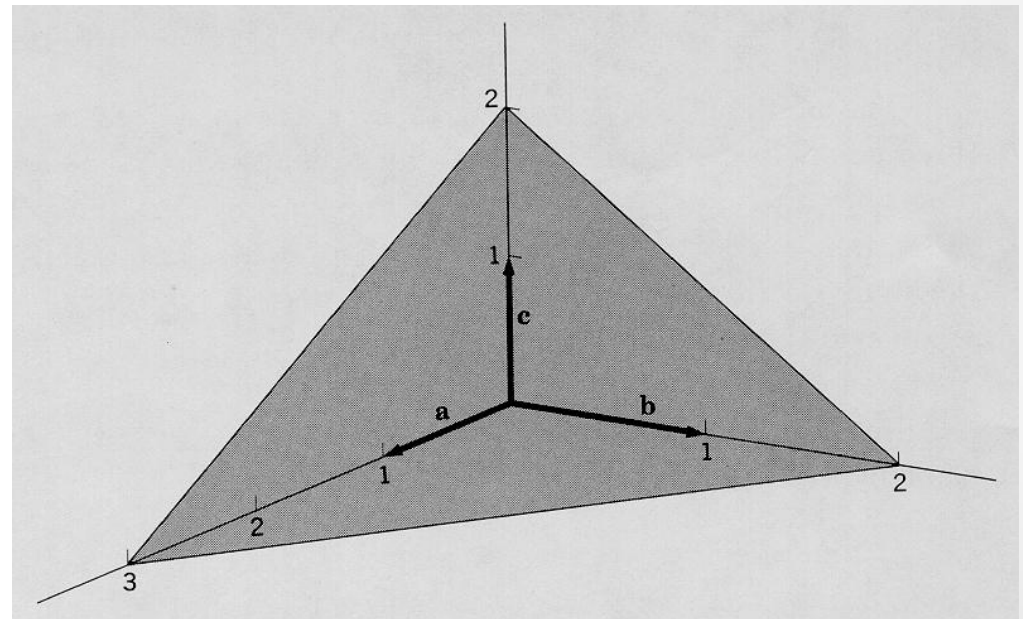
$$k = 1/2$$

$$l = 1/2$$



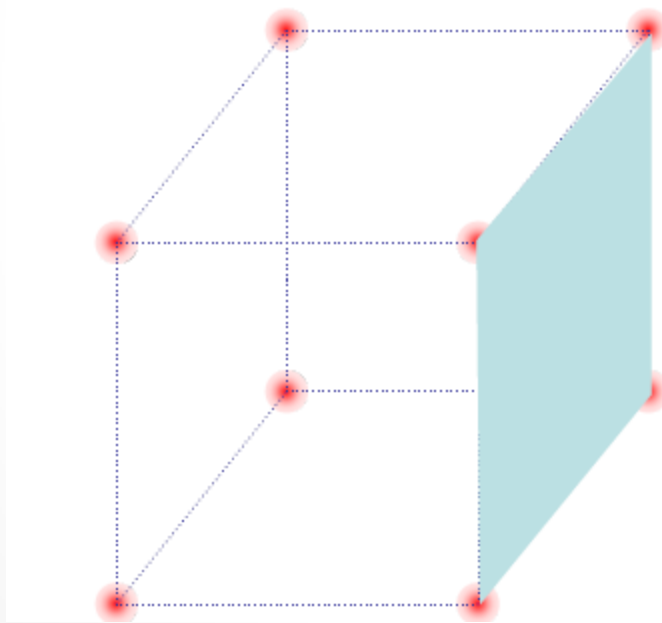
# Realization of Indices

- Find smallest integers with same ratio
  - $h = 1/3, k = 1/2, l = 1/2$
  - $h \rightarrow 2, k \rightarrow 3, l \rightarrow 3$
- Record index
  - $(hkl) = (233)$



# Planes by Symmetry

- Planes by symmetry are a 'family'
- Therefore,  $\{100\}$  consists of  $(100)$ ,  $(010)$ ,  $(001)$



For an isotropic material,  $(100)$  is only defined by reference axis



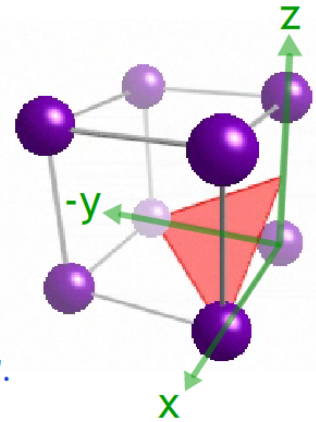
# Negative Indices

- Leave sign of negative intercept
- Find ratio as normal
- Denote negative with 'bar'

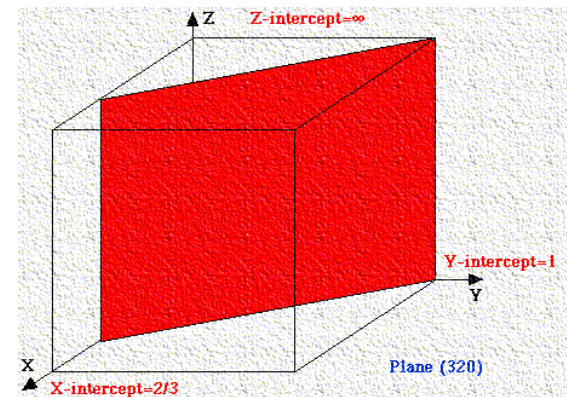
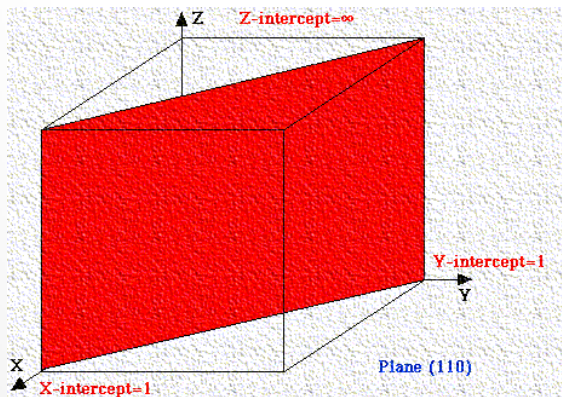
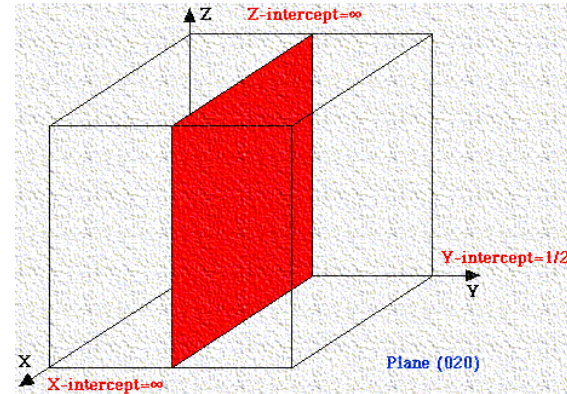
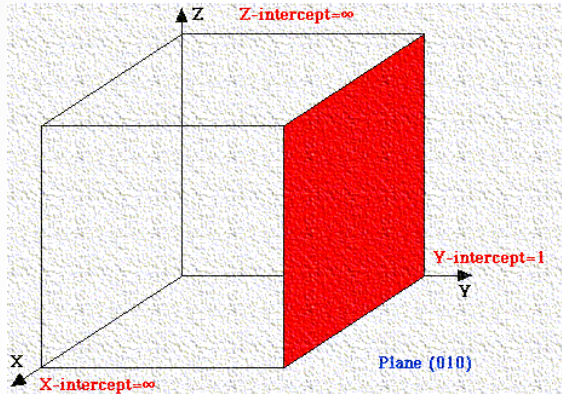
Instead of using a minus sign, we use an "overbar" to denote a negative index. This gives a Miller index of:

$$(1\bar{1}2)$$

This reads as: "one, bar one, two".



# Example of Indices



# Applications

- Determine material properties
- Semi-conductor properties depend on structure
- Structure affects energy band-gap
- Band-gap determines wavelengths absorbed/emitted
- Manipulate lattice structure (doping)

# References

- CRYSTAL STRUCTURE PART II. (n.d.).<http://www.ocf.berkeley.edu/>. Retrieved September 13, 2011, from [www.ocf.berkeley.edu/~mwg/lab/Miller.pdf](http://www.ocf.berkeley.edu/~mwg/lab/Miller.pdf)
- DoITPoMS - TLP Library Lattice Planes and Miller Indices. (n.d.). *Dissemination of IT for the Promotion of Materials Science (DoITPoMS)*. Retrieved September 13, 2011, from [http://www.doitpoms.ac.uk/tlplib/miller\\_indices/index.php](http://www.doitpoms.ac.uk/tlplib/miller_indices/index.php)
- Ferry, D., Vasieska, D., & Klimeck, G. (n.d.). Crystal Directions, Wave Propagation and Miller Indices. *nanohub*. Retrieved September 13, 2011, from [nanohub.org/resources/9137/download](http://nanohub.org/resources/9137/download)
- Solids / Crystal Structure and Solution. (n.d.).[Chemconnections.org](http://Chemconnections.org). Retrieved September 13, 2011, from [chemconnections.org/crystals/pdf/crys-solids.pdf](http://chemconnections.org/crystals/pdf/crys-solids.pdf)
- Streetman, B. G., & Banerjee, S. K. (2006). *Solid State Electronic Devices* (6a. ed.). Upper Saddle River, New Jersey: Prentice-Hall.